## LONG-TERM THERMAL STABILITY OF POLYIMIDE RESIN TRIA-X AND CARBO FIBER /TRIA-X COMPOSITE MATERIAL

Yuki Kubota<sup>1</sup>, Takefumi Furuta<sup>2</sup>, Yuichi Ishida<sup>1</sup>, Rikio Yokota<sup>2</sup>, Masahiko Miyauchi<sup>3</sup>, Toshio Ogasawara<sup>4</sup> and Takuya Aoki<sup>1</sup>

<sup>1</sup>Aeronautical Technology Directorate, Japan Aerospace Exploration Agency, 6-13-1 Ohsawa, Mitaka, Tokyo, 181-0015, Japan

Email: kubota.yuuki@jaxa.jp, Web Page: http://global.jaxa.jp/

<sup>2</sup>Frotier Materials Development Laboratories, KANEKA COPORATION, 2-3-18 Nakanoshima, Kitaku, Ohsaka, Japan

<sup>3</sup> Kaneka US Material Research Center, KANEKA Americas Holding., Inc., 800 Raymond Stotzer Pkwy, Suite #2105, 4478 TAMU, College Station, Texas, United States

<sup>4</sup>The graduate school of Engineering, Department of Mechanical System Engineering, Tokyo University of Agriculture and Technology, 2-24-16 Nakamachi, Koganei, Tokyo, Japan

Keywords: Aromatic polyimide, Polymer matrix composite, Long-term thermal stability, Oxidation.

#### Abstract

Long-term thermal stability of polyimide resin "TriA-X" and carbon fiber reinforced TriA-X was evaluated. The neat TriA-X resin was exposed at 180 °C, 240 °C, and 270°C for 3000 h in air and at 270°C under reduced-pressure environment. The composite material was exposed at 240 °C and 270 °C for 3000 h in air. The neat resin showed excellent thermal stability at 180°C for 3000h. Although the weight loss and shrinkage of the resin at 270°C was significant because of oxidation, the composite material maintained the compression strength after the aging at 270 °C for 2000h. The degradation of TriA-X resin in air was mainly dominated by the oxidation of 4-phenylethynylphtnalic anhydride moiety which was the end group of TriA-X.

## 1. Introduction

Aromatic polyimide resins have excellent potential for matrix of composite materials applied in high temperature aplications. NASA Langley Reserch Center had developed PMR-15 [1-4] showing a high glass transition temperature of 340 °C. However, it is brittle and shows low oxidation resistance. Although another developed polyimide resin PETI-5 has high fracture toughness, the glass transition temperature is 80 °C lower than that of PMR-15 [5-8].

In order to combine high glass transition temperature, good mechanical properties, and formability to polymer matrix composite, Yokota et al. developed an amorphous, asymmetric, and addition-type imide oligomer "Triple-A" [9, 10]. The chemical structure of Triple-A polyimide had been revised and improved several times. The most advanced polyimide of Triple-A series, "TriA-X", shows the world highest properties [11, 12]. The imide oligomer of TriA-X is derived from four monomers; pyromellitic dianhydride (PMDA), 2-phenyl-4, 4'-diaminodiphenyl ether (p-ODA), 9,9-bis (4-aminophenyl) fluorene (BAFL), and 4-phenylethynylphthalic anhydride (PEPA). Figure 1 shows the chemical structure of TriA-X imide oligomer. This imide oligomer shows low melt viscosity because of asymmetric structure ( $|\eta^*|_{\rm min}=154~{\rm Pa}\cdot{\rm s}$ ). Furthermore, the cured resin has high glass transition temperature ( $T_{\rm g}=370~{\rm °C}$ ) and excellent failure strain (>11%). The developed carbon fiber/TriA-X composite materials exhibits excellent mechanical properties at temperatures higher than 200 °C.

When the high temperature polymer matrix composite is applied for aerospace applications, such as a commercial aircraft engine and a leading edge of supersonic aircraft, knowledge of long-term thermal stability is an essential information to predict its lifetime and durability. In this study, the neat TriA-X resin was exposed at 180, 240, and 270°C for 3000 h in air so as to understand the potential of TriA-X for long-term thermal stability and the degradation process in air. Furthermore, the stability of carbon fiber/TriA-X composite material was evaluated by a non-hole compression test after aging at 240 °C and 270 °C for 2100 h in air.

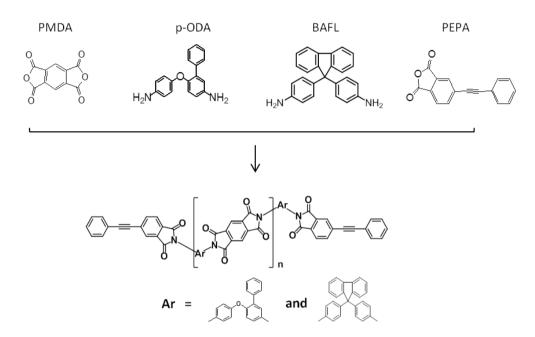


Figure 1. Chemical structure of the PMDA/p-ODA;BAFL/PEPA (TriA-X) imide oligomer

## 2. Experiments

## 2.1. Processing of the neat resin film

Three types of resin were prepeared for the isothermal aging test. Table 1 shows the molecular composition of these resins. TriA-X (B10) imide oligomer was prepeared by reacting two diamine monomers, p-ODA and BAFL, with one acid anhydride monomer: PMDA, and finally endcapping with PEPA. The molecular backbone of TriA-X, which was PMDA/p-ODA, was prepared so as to determine affects of the end group for the thermal stability. The imide oligomer was cured as a film (100mm × 100mm × 60~100µm) by using a hot press at 370°C for 1 hour under 1.4MPa. Furthermore, different TriA-X oligomers without BAFL (TriA-X (B0), n =1, 4, 7) in the degree of polymerization (*n*) were prepeared to evaluate effects of *n* for thermal stability. The forming of TriA-X (B0) film was more easily than that of TriA-X (B10) when *n* was high. This was the reason why the TriA-X (B0) was applied for the evaluation.

Table 1 Resin films for aging tests					
Specimen Type	Degree of	Ratio of each monomer			
	polymerization: n	p-ODA	BAFL	PMDA	PEPA
TriA-X(B10)	4	4.5	0.5	4.0	2.0
PMDA/p-ODA		5.0	0	5.0	0
	1	2.0	0	1.0	2.0
TriA-X(B0)	4	5.0	0	4.0	2.0
	7	8.0	0	7.0	2.0

#### 2.2. Fabricating of composite material

PAN-based carbon fiber (MR50R, Mitsubishi rayon co., Ltd.) was used as the reinforcement of composite. The fiber was immersed into the TriA-X(B10) imide oligomer by a prepreg coater. The prepregs then were stacked on a tool by hand lay-up. The stacking sequence was [90/0]4s. In this study, vacume hot pressureing (v-HP) molding was applied to fabricate the laminate plate. In v-HP molding, the prepreg was dired at 260 °C for 2 h, and cured at 370°C under 1.4MPa for 1 hour.

#### 2.3. Long-term thermal aging tests

The formed resin films were exposed at 180°C, 240°C, and 270°C for 3000 h in air using a thermostatic chamber (PHH-201M, ESPEC corp.), and at 270 °C for 3000 h under reduced-pressure environment using a vacuum furnace (DQ-30SA, SATO VAC INC.). The pressure in the vacuum furnace was less than 1.0 Pa. The film was cut to 50 mm  $\times$  100 mm  $\times$  60~100 µm as an exposed specimen. The weight loss and shrinkage during aging were measured. Their intial values were determined by the film dried at 60°C for more than 4 h under a reduced-pressure.

In the case of composite materials, the laminate was cut to 15 mm  $\times$  82 mm  $\times$  2.5 mm (w  $\times$  l  $\times$  t). Cut specimens were exposed at 240 °C and 270 °C in air using PHH-201M for 2100 h. According to the literature [13], it took 120 h at 120 °C to dry a polyimide composite. In this study, the weight exposed for 100 h at respective aging temperature was set as the initial value.

#### 2.4. Evaluation of mechanical properties of exposed composites

 $T_{\rm g}$  of the composite was measured by a dynamic mechanical analyzer (DMA: Q-800, TA Instruments). In this measurement, the single cantilever beam test mode was applied. The specimen size was 10 mm × 40 mm × 2.2 mm (w × 1 × t). The tests was run from 30 °C to 450 °C at 4.0 °C/min with a cycling rate of 1 Hz and strain of 0.1 %.

Non-hole compressive (NHC) strengths and elastic modulus were measured on the composite materials exposed for 1100 h, 2100 h, and 3100 h. In this measurement, NAL-II method [14] was applied. Figure 2 shows the jig used in NAL-II method. Lower and upper grips are fixed to a specimen (15mm  $\times$  82mm  $\times$  2.5mm) using four screw bolts. Compressive load is applied to a speceimen through the grips. Gauge length is 10 mm, therefore strain gauges can be used to measure simultaneously.

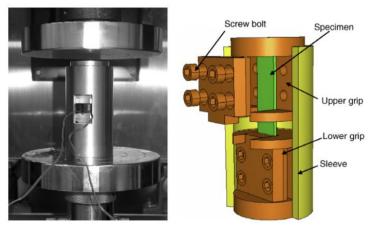


Figure 2 The jig used in NAL-II method [10, 14].

# 3. Reusult and discussion 3.1. TriA-X resin film

Figure 3 shows the weight loss of films as a function of the aging time. Furthermore, the shrinkage of the films were shown in Fig 4. As shown in Fig 3, the weight loss of TriA-X (B10) at  $180^{\circ}$ C was less than 0.5 %. The weight loss was significant with increase in temperature, which were 2.0 % at 240 °C and 8.9 % at 270 °C. This tendency agreed with the shrinkage as shown in Fig 4.

As shown in fig 3, the weight loss of TriA-X (B10) under reduced-pressure environment were negligible small compared with that in air. Furthermore, PMDA/p-ODA did not show apparent weight loss at 270 °C in air. These results indicated that the main reason of degradation of TriA-X(B10) was the oxidation on PEPA moiety which was the end group of TriA-X. Based on these results, the weight loss of TriA-X(B0) were compared on different degree of polymerization (n) as shown in Fig 5. As shown in Fig 5, the weight loss was small with increase in n. Since the molecular weight of the backbone (PMDA, p-ODA) becomes higher with increase in n, the concentration of PEPA moiety is relatively low when the n is high. Therefore, the result shown in Fig. 5 ensured that the oxidation of PEPA moiety was the main reason for the degradation of TriA-X.

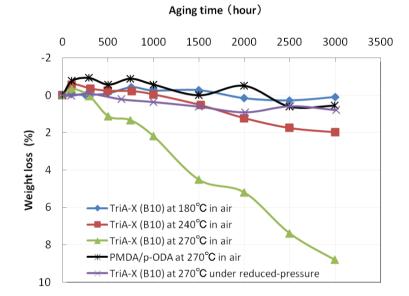


Figure 3 Weight loss of TriA-X (B10) and PMDA/p-ODA films exposed at 180°C , 240 °C, and 270 °C in air as a function of aging time.

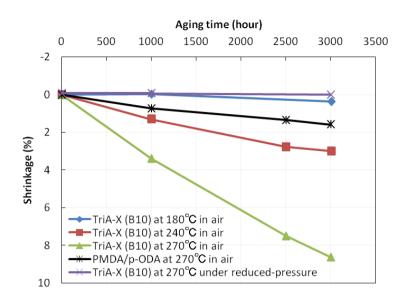


Figure 4 Shrinkage of surface area of TriA-X (B10) and PMDA/p-ODA films exposed at 180°C , 240 °C, and 270 °C in air as a function of aging time.

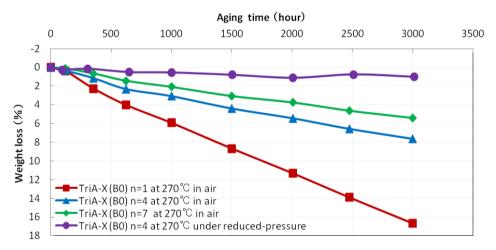


Figure 5 Weight loss of different TriA-X(B0) in the dgree of polymerizations (n).

## 3.2. MR50R/TriA-X

Figure 6 presents the a DMA trace of the MR50R/TriA-X(B10) composite. The storage modulus (*E'*), loss modulus (*E''*), and tan  $\delta$  are ploted as a function of temperature. The glass transition temperatue ( $T_g$ ) was determined from *E'* by setting up tangents to the linear portion of the curve before and after the modulus drop off. As a result, the  $T_g$  of the MR50R/TriA-X(B10) was 374°C that was close to that of the TriA-X (B10). This result implied that the matrix was totally cured.

Figure 7 shows the weight loss of the MR50R/TriA-X (B10) composite at 240 °C and 270 °C in air. Furthermore, Fig 8 and 9 presents the NHC strength and elastic modulus of the composite after aging at 240 °C and 270 °C. Although the weight of composite tended to decrease at 240 °C and 270 °C in air as well as the film, the compressive strength and elastic modulus were maintained at least for 2100 h at 270 °C in air. In general, the oxidation reaction of a composite material progresses from the surface. In this study, the surface of specimen was 90° ply which rarely beared the longtitudinally compression stress. Therefore, the reason why the compressive properties were maintained for 2100 h

at 270 °C might be that the degradation of composite progressed only at the surface ply.

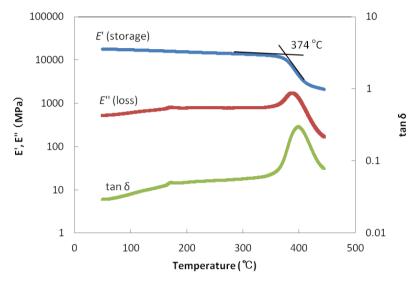


Figure 6 Dynamic mechanical snalysis (DMA) traces of a MR50R/TriA-X (B10) composite.

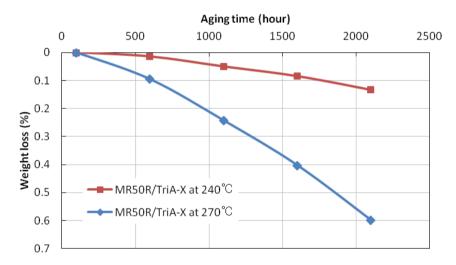


Figure 7 Weight loss of MR50R/TriA-X (B10) on aging tests at 240 °C and 270 °C in air.

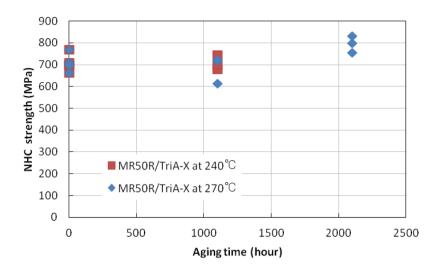


Figure 8 NHC strength of MR50R/TriA-X (B10) after aging tests at 240 °C and 270 °C in air.

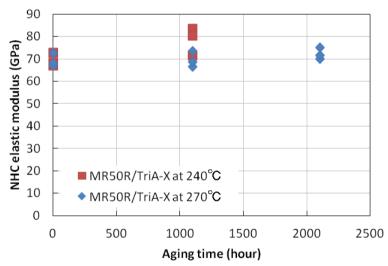


Figure 9 NHC elastic modulus of MR50R/TriA-X (B10) after aging tests at 240 °C and 270 °C in air.

## 4. Conclusions

This study evaluated the long-term thermal stability of a polyimide resin "TriA-X" at 180 °C, 240 °C and 270 °C. Furthermore, the thermal stability of carbon fiber/TriA-X composite material was evaluated by the non-hole compression test after aging at 240 °C and 270 °C. The TriA-X resin indicated high thermal stability at least for 3000h at 180 °C. The weight loss and shrinkage of TriA-X became significant at 270 °C. This degradation was mainly caused by the oxidation of the end group of TriA-X, which was 4-phenylethynylphtnalic anhydride moiety. Although the weight of composite decreased by the oxidation at 240 °C and 270 °C, the compression strength was maintained at 270 °C at least for 2100 h.

## Acknowledgement

This investigation have been financially supported by A03 unit in Structural Materials for Innovation in Cross-ministerial Strategic Innovation Promotion Program (SIP) from JST. Our deepest appreciation goes to all persons concerned in this program.

## References

- [1] T.T. Selafini. *Polyimides: Synthhesis Characterization, and Applications*, vol. 2 Mittal K. L., eds., New York: Plenum, p54, 1984.
- [2] R.H. Pater. Thermosetting polimides: a review. SAMPE. J. 30(5):113-122, 1994.
- [3] D. Wilson. PMR15-processing, properties and problems-a review. *British Polymer Jounal*, 20(5):405-416, 1988.
- [4] T.T. Serafini, P. Delvigs, and G. R. Lightsey. Thermally stable polyimides from solutions of monomeric reactions. *Journal of Applied Polymer Science*, 16:905-915, 1972
- [5] J. A. Hinkley and B. J. Jensen. The relative thermooxidative stability of a phthalimide against a phenylethynylphthalimide terminated polyimide. *High Performance Polymer*, 7:1-9, 1995.
- [6] T.H. Hou, B. J. Jensen, and P. M. Hergenrother, Processing and Properties of IM7/PETI Composites. *Journal of Composite Material*, 30:109-122, 1996.
- [7] P.M. Hergenrother. Development of composites, adhesives and sealants for high-speed commercial airplanes. *SAMP Japan*, 36(1): 113-122, 1994.
- [8] P.M. Hergenrother. Resin transfer molding and resin infusion fabrication of high temperature composites. *SAMPE Japan*, 36(3):32-41, 2000.
- [9] R. Yokota, S. Yamamoto, S. Yano, T. Sawaguchi, M. Hasegawa, H. Yamaguchi, H. Ozawa and R. Sato, Molecular desining of heat resistant polyimides having excellent processability and high glass transition temperature *High Performance Polymer.*, 13 pp.S61–72 (2001).
- [10] T. Ogasawara, Y. Ishida, R. Yokota, T. Watanabe, T. Aoi, and J. Goto. Processing and properties of carbon fiber / Triple-A polyimide composites fabricated from imide oligomer dry prepreg *Composite Part A*, 38:1296-1303, 2007.
- [11] M. Miyauchi, Y. Ishida, T. Ogasawara and R. Yokota, Novel phenylethynyl-terminated PMDAtype polyimides based on KAPTON backbone structures derived from 2-phenyl-4,4'diaminodiphenyl ether. *Polymer Journal*, 44:959-965, 2012.
- [12] Y. Ishida, M. Miyauchi, T. Ogasawara, and R. Yokota. Development of "TriA-X" polyimide/carbon fiber composites prepared by imide solution prepregs. *Proceedings of the 18th International Conference on Composite Materials ICCM-18, Jeju, Korea, August 21-26 2011.*
- [13] T. Shimokawa, Y. Hamaguchi, and H. Katoh. Effect of moisture absorption on hot/wet compressive strength of T800H/PMR-15 carbon/polyimide. *Journal of Composites Materials* 33(18):1685-1698, 1999.
- [14] T. Ogasawara and T. Ishikawa. Evaluation of standard compressive test methods for cabon fiber composites and proposal of a simple test method (NAL-II), JAXA Research and Development Memorandum, ISSN 1349-1121, 2009